

Preferred Solution Conformation of Peptides Rich in the Lipophilic, Chiral, C^α-Methylated α-Amino Acid (αMe)Aoc

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Abstract: The lipophilic, chiral, C^α-methylated α-amino acid L-(αMe)Aoc (2-methyl-2-amino-octanoic acid) was prepared using a chemo-enzymatic approach. Two series of terminally protected model peptides, from dimer through to hexamer, containing L-(αMe)Aoc in combination with either Gly or Aib, were synthesized by solution methods and were fully characterized. A solution conformational analysis, based on FT-IR absorption, ¹H-NMR and circular dichroism (CD) techniques, was performed with the aim at determining the preferred conformation of this novel amino acid and the relationship between chirality at its α-carbon atom and screw sense of the helix that is formed. The results obtained strongly support the view that L-(αMe)Aoc favours the formation of the *right*-handed 3₁₀-helical conformation. Copyright © 1999 European Peptide Society and John Wiley & Sons, Ltd.

Keywords: α-aminoisobutyric acid; 3₁₀-helix; hydrophobicity; lipophilic side-chain; C^α-methyl, C^α-*n*-hexylglycine; peptide conformation; peptide synthesis; spectroscopy

INTRODUCTION

Interest in chiral, C^α-methylated, α-amino acids stems from their capability to induce significant restraints on peptide backbone conformational freedom [1,2]. In particular, these building blocks have gained importance in the design of analogues of bioactive compounds [3–11]. The conformational propensity of C^α-methylated analogues of protein

amino acids, such as (αMe)Val [2,12,13], (αMe)Leu [2], (αMe)Phe [2] and (αMe)Trp [14], when introduced in peptide sequences, have been extensively investigated.

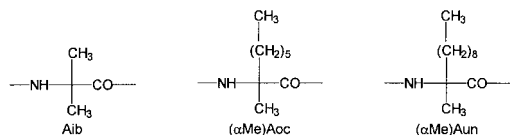
The possibility of combining the conformational rigidity that C^α-methylation induces in the peptide backbone with the specific properties of the amino acid side-chains has attracted the attention of chemists towards the production of an increasing number of compounds of this class. In particular, as lipophilicity plays an important role in improving peptide drug bioavailability, C^α-*un*methylated lipoamino acids have been frequently introduced into peptides to increase their overall hydrophobicity and to impart to them a membrane-like character [15–20]. In this way, the passage of poorly absorbed drugs across biological membranes is increasingly facilitated. Interestingly, the combination of the atypical nature of a lipoamino acid with the even more atypical nature of a C^α-methylated amino

Abbreviations: Aib, α-aminoisobutyric acid or C^α-dimethylglycine; (αMe)Aoc, C^α-methyl, C^α-*n*-hexylglycine or 2-methyl-2-amino-octanoic acid; (αMe)Aun, C^α-methyl, C^α-*n*-nonylglycine or 2-methyl-2-amino-undecanoic acid; DMSO, dimethylsulphoxide; EDC, *N*-ethyl, *N'*-[3-(dimethylamino)propyl]carbodiimide; HOAt, 1-hydroxy-7-azabenzotriazole; NMM, *N*-methylmorpholine; Z, benzoyloxycarbonyl; MeCN, acetonitrile; MeOH, methanol; TEMPO, 2,2,6,6-tetramethylpiperidiny-1-oxy; TFE, 2,2,2-trifluoroethanol.

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acid might also contribute to imparting a high degree of resistance to proteolytic enzymes, known to be responsible for the chemical instability of peptide drugs. In this context, recently, the highly lipophilic, chiral, C^α-methylated α-amino acid L-(αMe)Aun has been introduced in an analogue of the membrane-active lipopeptaibol antibiotic trichogin GA IV, where it successfully replaced the naturally occurring N-terminal n-octanoyl chain, essential for activity [21].



As no information is available to date on the conformational propensity of lipophilic, chiral, C^α-methylated α-amino acids, this paper presents the synthesis, characterization and conformational analysis [by FT-IR absorption, NMR and circular dichroism (CD) techniques] of a number of model peptides containing L-(αMe)Aoc in combination with Aib or Gly.

MATERIALS AND METHODS

Peptide Synthesis

Melting points were determined using a Leitz (Wetzlar, Germany) model Laborlux 12 apparatus and were not corrected. Optical rotations were measured using a Perkin-Elmer (Norwalk, CT) model 241 polarimeter equipped with a Haake (Karlsruhe, Germany) model D thermostat. Thin layer chromatography (TLC) was performed on Merck (Darmstadt, Germany) Kieselgel 60/F₂₅₄ precoated plates. The chromatograms were developed by quenching of UV fluorescence, chlorine-starch-potassium iodide or ninhydrin chromatic reaction as appropriate.

Infrared Absorption

The solid state infrared absorption spectra (KBr disk technique) were recorded with a Perkin-Elmer model 580 B spectrophotometer equipped with a Perkin-Elmer model 3600 IR data station. The solution IR absorption spectra were recorded using a Perkin-Elmer model 1720 X FT-IR spectrophotometer, nitrogen-flushed, equipped with a sample shuttle device, at 2 cm⁻¹ nominal resolution, averaging 100 scans. Solvent (baseline) spectra were obtained under the same conditions. Cells with path lengths

of 0.1, 1.0 and 10 mm (with CaF₂ windows) were used. Spectrograde deuteriochloroform (99.8% d) was purchased from Fluka (Buchs, Switzerland).

Circular Dichroism

The CD spectra were obtained on a Jasco (Tokyo, Japan) model J-715 dichrograph. Cylindrical fused quartz cells of 1.0 and 0.2 mm path lengths were used. The values are expressed in terms of $[\theta]_T$, the total molar ellipticity (deg cm² dmol⁻¹). Spectrograde MeOH and TFE (Acros Organics, Geel, Belgium) were used as solvents.

¹H-Nuclear Magnetic Resonance

The ¹H-NMR spectra were recorded with a Bruker (Karlsruhe, Germany) model AM 400 spectrometer. Measurements were carried out in deuteriochloroform (99.96% d; Aldrich, Milwaukee, WI) and deuterated dimethylsulphoxide (99.96% d₆; Acros Organics, Geel, Belgium) with tetramethylsilane as the internal standard. The free radical TEMPO was purchased from Sigma (St. Louis, MO).

RESULTS AND DISCUSSION

Peptide Synthesis

For the large-scale production of the enantiomerically pure L-(αMe)Aoc, an economically attractive and generally applicable chemo-enzymatic synthesis, developed by DSM Research [22–24] a few years ago, was exploited. It involves a combination of partial Strecker synthesis for the preparation of the racemic α-amino acid amide followed by the use of a broadly specific amino acid amidase from *Mycobacterium neoaurum* to achieve optical resolution. The synthesis of racemic (αMe)Aoc and the gas chromatographic separation of selected diastereomeric esters at the analytical level have already been published [25]. The Z-protected derivative was prepared by reacting the free amino acid with N-(benzyloxycarbonyl)-succinimide in MeCN in the presence of the lipophilic base tetramethylammonium hydroxide [26]. Gly and Aib *tert*-butyl esters were obtained by esterification of the corresponding Z-protected amino acid with isobutylene in the presence of a catalytic amount of sulphuric acid [27]. Peptide bond formation was achieved by the EDC/HOAt method [28] in CH₂Cl₂ in the presence of NMM. Using this approach, formation of the sterically hindered (αMe)Aoc-Aib and Aib-(αMe)Aoc peptide

bonds was obtained in moderately good yields (55–70%). Removal of the *Z* *N*-protecting group was performed by catalytic hydrogenation. The *N*^z-blocked hexapeptide Ac-Aib-L-(αMe)Aoc-Aib-Aib-L-(αMe)Aoc-Aib-OtBu was prepared by treatment of H-Aib-L-(αMe)Aoc-Aib-Aib-L-(αMe)Aoc-Aib-OtBu with ten equivalents of acetic anhydride in CH_2Cl_2 . The *N*^z-protected peptide free acids were prepared from the corresponding *tert*-butyl esters by treatment with diluted trifluoroacetic acid. The physical and analytical properties of the (αMe)Aoc derivatives and peptides are listed in Table 1. All compounds were also characterized by ¹H-NMR.

Solution Conformational Analysis

The solution conformational preferences of the (αMe)Aoc/Aib and (αMe)Aoc/Gly model peptides were determined by FT-IR absorption, ¹H-NMR and CD techniques.

Figure 1 shows the FT-IR absorption spectra (N–H stretching region) of the (αMe)Aoc/Aib series from di to hexapeptide in CDCl_3 solution. The curves of the higher homologues are characterized by two bands at about 3430 cm^{-1} , assigned to the free (solvated) NH groups, and at $3380\text{--}3340\text{ cm}^{-1}$, assigned to H-bonded NH groups [29]. The intensity of the low-frequency band, relative to the high-frequency

band, considerably increases as the main chain length increases. Concomitantly, the absorption maximum markedly shifts to lower wavenumbers. Figure 2 illustrates the FT-IR absorption spectra of the (αMe)Aoc/Gly series from di to pentapeptide in the same spectral region. A behaviour similar to that exhibited by the (αMe)Aoc/Aib series, albeit less pronounced, is observed. In the (αMe)Aoc/Gly di, tri and tetrapeptides, a significant amount of un-ordered conformation is still present, together with a modest amount of H-bonded conformations. Only in the pentapeptide, with two (αMe)Aoc residues, a significant shift to a lower frequency and a remarkable increase in intensity of the absorption band of the H-bonded NH groups, indicative of the onset of a stable helical structure, are seen. Upon changing the concentration (in the 10–0.1 mM range) all model peptides display only minor variations in the spectra (results not shown). The present FT-IR absorption investigation provided convincing evidence that the (αMe)Aoc/Aib peptides are highly folded in a helical conformation, whereas a significant flexibility seems to be characteristic of the lower members of the (αMe)Aoc/Gly series.

To get more detailed information on the preferred conformation of the (αMe)Aoc-rich peptides, a ¹H-NMR study in a CDCl_3 solution at 1.0 mM concentration was performed, where self-association is

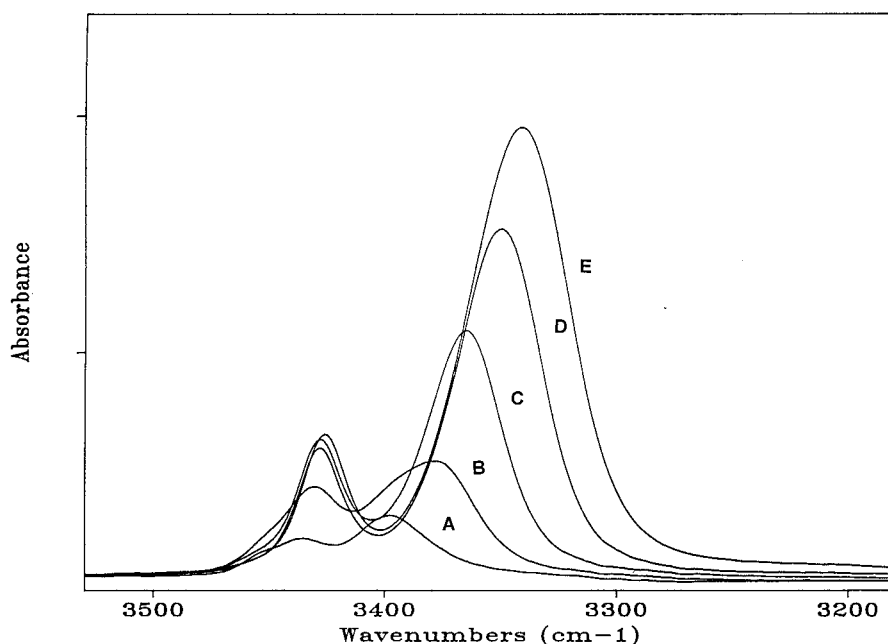


Figure 1 FT-IR absorption spectra in the $3500\text{--}3200\text{ cm}^{-1}$ region of Z-L-(αMe)Aoc-Aib-OtBu (A), Z-Aib-L-(αMe)Aoc-Aib-OtBu (B), Z-Aib-Aib-L-(αMe)Aoc-Aib-OtBu (C), Z-L-(αMe)Aoc-Aib-Aib-L-(αMe)Aoc-Aib-OtBu (D), and Z-Aib-L-(αMe)Aoc-Aib-Aib-L-(αMe)Aoc-Aib-OtBu (E) in CDCl_3 solution. Peptide concentration: 1.0 mM.

Table 1 Physical and Analytical Properties for the (α Me)Aoc Derivatives and Peptides

Compound	Yield (%)	Melting point (°C)	Recrystallization solvent ^a	[α] _D ²⁰ (°) ^b	TLC ^c			IR ^d (cm ⁻¹)
					R _F I	R _F II	R _F III	
Z-L-(α Me)Aoc-OH	90	Oil	—	3.6	0.70	0.90	0.30	3414, 3333, 1706, 1585
Z-L-(α Me)Aoc-Aib-OtBu	70	89–91	EtOAc/PE	2.0	0.90	0.95	0.60	3404, 3294, 1722, 1658, 1537
Z-L-(α Me)Aoc-Aib-OH	96	128–130	Et ₂ O/PE	1.1	0.45	0.95	0.20	3427, 3293, 1719, 1703, 1655, 1590, 1529
Z-Aib-L-(α Me)Aoc-Aib-OtBu	55	Oil	—	–6.0	0.90	0.95	0.40	3432, 3350, 1731, 1704, 1526
Z-Aib-L-(α Me)Aoc-Aib-OH	70	141–143	Et ₂ O/PE	–1.8	0.40	0.95	0.20	3314, 1735, 1698, 1659, 1526
Z-Aib-Aib-L-(α Me)Aoc-Aib-OtBu	65	158–160	EtOAc/PE	–17.0	0.80	0.95	0.25	3425, 3335, 1728, 1683, 1668, 1534
Z-Aib-Aib-L-(α Me)Aoc-Aib-OH	75	Oil	—	–5.5	0.35	0.95	0.15	3314, 1703, 1660, 1529
Z-L-(α Me)Aoc-Aib-Aib-L-(α Me)Aoc-Aib-OtBu	60	205–206	EtOAc/PE	–5.6	0.90	0.95	0.25	3417, 3322, 1724, 1699, 1663, 1532
Z-Aib-L-(α Me)Aoc-Aib-Aib-L-(α Me)-Aoc-Aib-OtBu	61	146–147	EtOAc/PE	–12.0	0.80	0.95	0.35	3427, 3322, 1730, 1701, 1661, 1530
Ac-Aib-L-(α Me)Aoc-Aib-Aib-L-(α Me)-Aoc-Aib-OtBu	70	85–86	CH ₂ Cl ₂ /PE	–3.5	0.40	0.90	0.15	3307, 1730, 1659, 1535
Z-L-(α Me)Aoc-Gly-OtBu	97	Oil	—	2.6	0.90	0.90	0.65	3374, 1726, 1660, 1584
Z-Gly-L-(α Me)Aoc-Gly-OtBu	80	114–115	EtOAc/PE	1.6	0.90	0.95	0.30	3353, 3275, 1730, 1701, 1677, 1639, 1552
Z-Gly-Gly-L-(α Me)Aoc-Gly-OtBu	87	125–126	EtOAc/PE	–0.8	0.65	0.95	0.20	3339, 1745, 1715, 1663, 1524
Z-L-(α Me)Aoc-Gly-Gly-L-(α Me)Aoc-Gly-OtBu	88	59–60	Et ₂ O/PE	–1.9	0.65	0.95	0.20	3315, 1744, 1660, 1533

^a EtOAc, ethyl acetate; PE petroleum ether; Et₂O, diethyl ether.

^b $c = 0.5$, MeOH.

^c R_FI = CHCl₃/EtOH, 9:1; R_FII = 1-BuOH/AcOH/H₂O, 3:1:1; R_FIII = toluene/EtOH, 7:1.

^d The IR absorption spectra were obtained in KBr pellets (only bands in the 3500–3200 and 1800–1500 cm⁻¹ regions are reported).

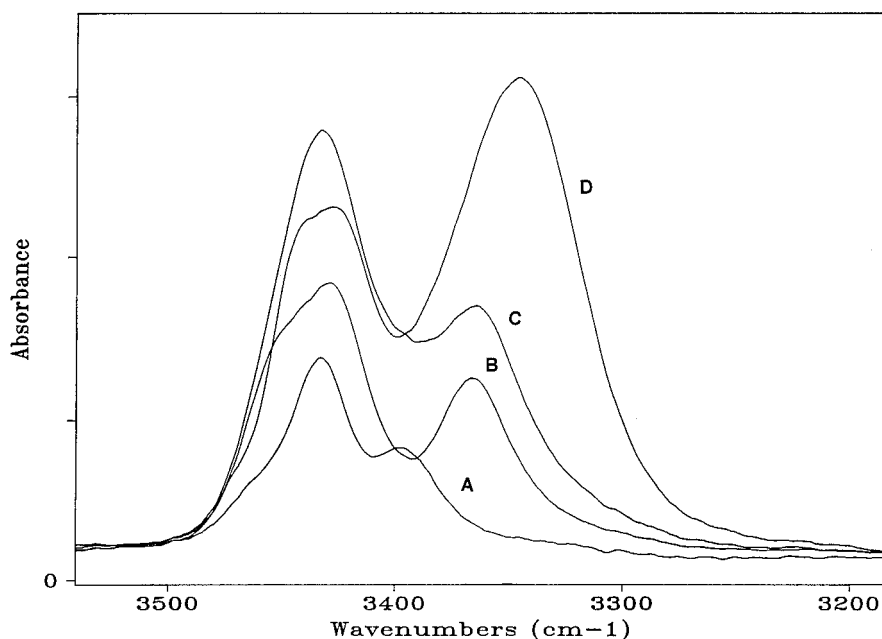


Figure 2 FT-IR absorption spectra in the 3500–3200 cm^{-1} region of Z-L-(α Me)Aoc-Gly-OtBu (A), Z-Gly-L-(α Me)Aoc-Gly-OtBu (B), Z-Gly-Gly-L-(α Me)Aoc-Gly-OtBu (C), and Z-L-(α Me)Aoc-Gly-Gly-L-(α Me)Aoc-Gly-OtBu (D) in CDCl_3 solution. Peptide concentration: 1.0 mM.

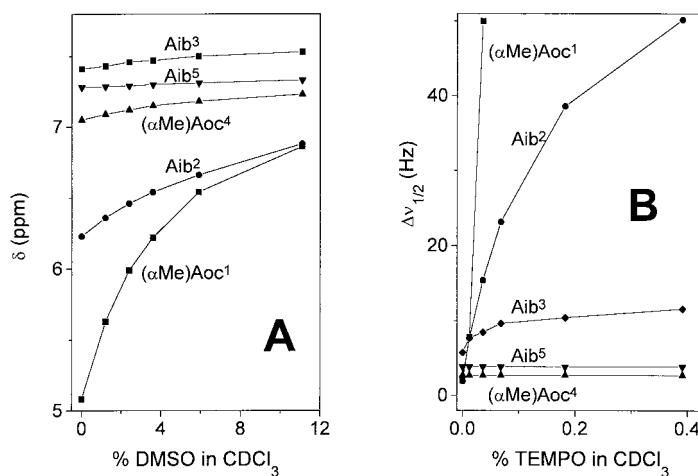


Figure 3 $^1\text{H-NMR}$ titration of Z-L-(α Me)Aoc-Aib-Aib-L-(α Me)Aoc-Aib-OtBu: (A) plot of NH chemical shifts as a function of increasing percentages of DMSO added to the CDCl_3 solution (v/v); (B) plot of bandwidth of the NH signals as a function of increasing percentages of TEMPO (w/v) in CDCl_3 . Peptide concentration: 1.0 mM.

absent. The delineation of inaccessible (intramolecularly H-bonded) NH groups was carried out by analysing the behaviour of the NH resonances upon addition of two perturbing agents. In particular, (i) the solvent dependence of NH chemical shifts was examined by adding increasing amounts of the strong H-bonding acceptor solvent DMSO [30,31] to the CDCl_3 solution, and (ii) the line broadening of

NH resonances induced by adding the free radical TEMPO was also examined [32].

Figure 3 illustrates the behaviour of the NH resonances of the (α Me)Aoc/Aib pentapeptide upon addition of DMSO and TEMPO. Figure 4 shows the behaviour of the related (α Me)Aoc/Gly pentapeptide under the same conditions. All -CONH- proton resonances were assigned by means of 2D ROESY

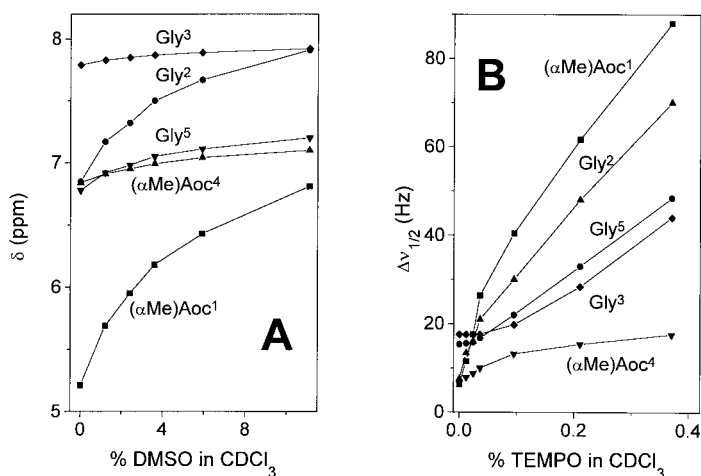


Figure 4 $^1\text{H-NMR}$ titration of Z-L-(α Me)Aoc-Gly-Gly-L-(α Me)Aoc-Gly-OtBu: (A) plot of NH chemical shifts as a function of increasing percentages of DMSO added to the CDCl_3 solution (v/v); (B) plot of bandwidth of the NH signals as a function of increasing percentages of TEMPO (w/v) in CDCl_3 . Peptide concentration: 1.0 mM.

experiments. In each peptide two classes of NH protons were found. Class (i) [N(1)H and N(2)H protons] includes protons whose chemical shifts are remarkably sensitive to the addition of DMSO and whose resonances broaden significantly upon addition of the paramagnetic perturbing agent TEMPO. Class (ii) [N(3)H, N(4)H and N(5)H protons] includes those displaying a behaviour characteristic of shielded protons (relative insensitivity of the chemical shift to solvent composition and of the line width to the presence of TEMPO).

To summarize, these $^1\text{H-NMR}$ results allow us to conclude that, in a CDCl_3 solution, the N(3)–N(5)H protons of the two pentapeptides are almost inaccessible to perturbing agents and are therefore most probably intramolecularly H-bonded. Therefore, it is reasonable that the most populated conformation adopted in the CDCl_3 solution by the terminally protected (α Me)Aoc-rich pentapeptides would be the 3_{10} -helix, where only the two N-terminal NH protons do not participate in the intramolecular H-bonding scheme. However, the N(3)H and N(5)H protons of the pentapeptide Z-(α Me)Aoc-(Gly)₂-(α Me)Aoc-Gly-OtBu exhibit a higher sensitivity to the addition of TEMPO as compared with that of the corresponding protons of Z-(α Me)Aoc-(Aib)₂-(α Me)Aoc-Aib-OtBu. In agreement with the FT-IR absorption results, this observation points to a higher degree of conformational flexibility in the (α Me)Aoc/Gly peptides.

The hexapeptide Ac-Aib-L-(α Me)Aoc-(Aib)₂-L-(α Me)Aoc-Aib-OtBu, lacking any potentially disturbing chromophoric group at the N-terminus, was

synthesized with the aim of determining by CD the relationship between chirality at the α -carbon atom of this new α -amino acid and the screw sense of the helix that is formed. The CD spectrum of this (L,L)-peptide in MeOH solution (Figure 5), displaying a negative maximum centred at 203 nm and a weak, negative shoulder at about 220 nm, strongly resembles the dichroic pattern canonical for a right-handed 3_{10} -helix [33]. This type of helical structure appears to be rather stable, as no relevant changes are observed by changing the solvent polarity. From

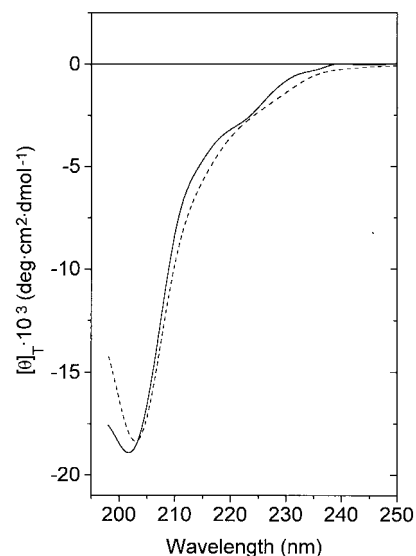


Figure 5 CD spectra of Ac-Aib-L-(α Me)Aoc-Aib-Aib-L-(α Me)Aoc-Aib-OtBu in MeOH (solid line) and TFE (dashed line) solutions. Peptide concentration: 0.5 mM.

this study the additional information can be extracted that the screw sense of the helix formed by this C $^{\alpha}$ -methylated α -amino acid is the same as that characteristic of protein amino acids (an L-amino acid gives a *right*-handed helix).

CONCLUSIONS

The experimental results reported in this work have contributed to elucidate the conformational preferences of the lipophilic, chiral, C $^{\alpha}$ -methylated α -amino acid (α Me)Aoc. The FT-IR absorption, CD and NMR analyses agree with a highly populated 3_{10} -helical motif for the longer model peptides in solution. Moreover, we have been able to show that L-(α Me)Aoc induces the formation of a *right*-handed helical structure. Finally, a study of the dependence of the preferred conformation in chloroform solution on peptide concentration did not reveal any aggregation phenomenon driven by the side-chain of this residue. The authors foresee a bright future for the new class of strongly helicogenic and highly lipophilic α -amino acids, of which (α Me)Aoc and (α Me)Aun [21] are significant members.

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